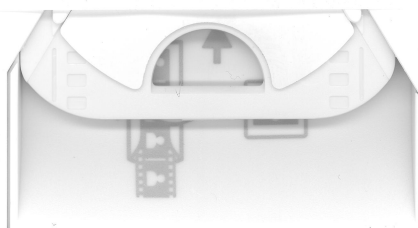


Identifying Element Symbols and Common Ions

Name Key Period _____ Date _____

Name	Common Ion (Na^{+1})	Cation or Anion	Anion "ide" name
Strontium	Sr^{+2}	C	—
Iodine	I^{-1}	A	Iodide
Nitrogen	N^{-3}	A	Nitride
Oxygen	O^{-2}	A	Oxide
Lithium	Li^{+1}	C	—
Magnesium	Mg^{+2}	C	—
Silver	Ag^{+1}	C	—
Zinc	Zn^{+2}	C	—
Iron (II)	Fe^{+2}	C	—
Phosphorous	P^{-3}	A	Phosphide
Sulfur	S^{-2}	A	Sulfide
Aluminum	Al^{+3}	C	—
Fluorine	F^{-1}	A	Fluoride
Chlorine	Cl^{-1}	A	Chloride
Potassium	K^{+1}	C	—
Bromine	Br^{-1}	A	Bromide
Cesium	Cs^{+1}	C	—
Francium	Fr^{+1}	C	—
Chromium (VI)	Cr^{+6}	C	—



Lead (IV)	Pb^{+4}	C	—
Vanadium (V)	V^{+5}	C	—
Iron (III)	Fe^{+3}	C	—
Cobalt (II)	Co^{+2}	C	—
Lead (II)	Pb^{+2}	C	—
Tin (IV)	Sn^{+4}	C	—
Cobalt (III)	Co^{+3}	C	—
Tin (II)	Sn^{+2}	C	—
Arsenic	As^{-3}	A	Arsenide
Nickel (II)	Ni^{+2}	C	—
Gallium	Ga^{+3}	C	—
Antimony (IV)	Sb^{+4}	C	—
Sodium	Na^{+1}	C	—
Barium	Ba^{+2}	C	—
Boron	B^{+3}	C	—
Beryllium	Be^{+2}	C	—
Manganese (II)	Mn^{+2}	C	—
Chromium (III)	Cr^{+3}	C	—
Gold (II)	Au^{+2}	C	—

